"Spent" fuel from a typical (old) reactor:

What's in it?

- What's in it?
- What does it do?

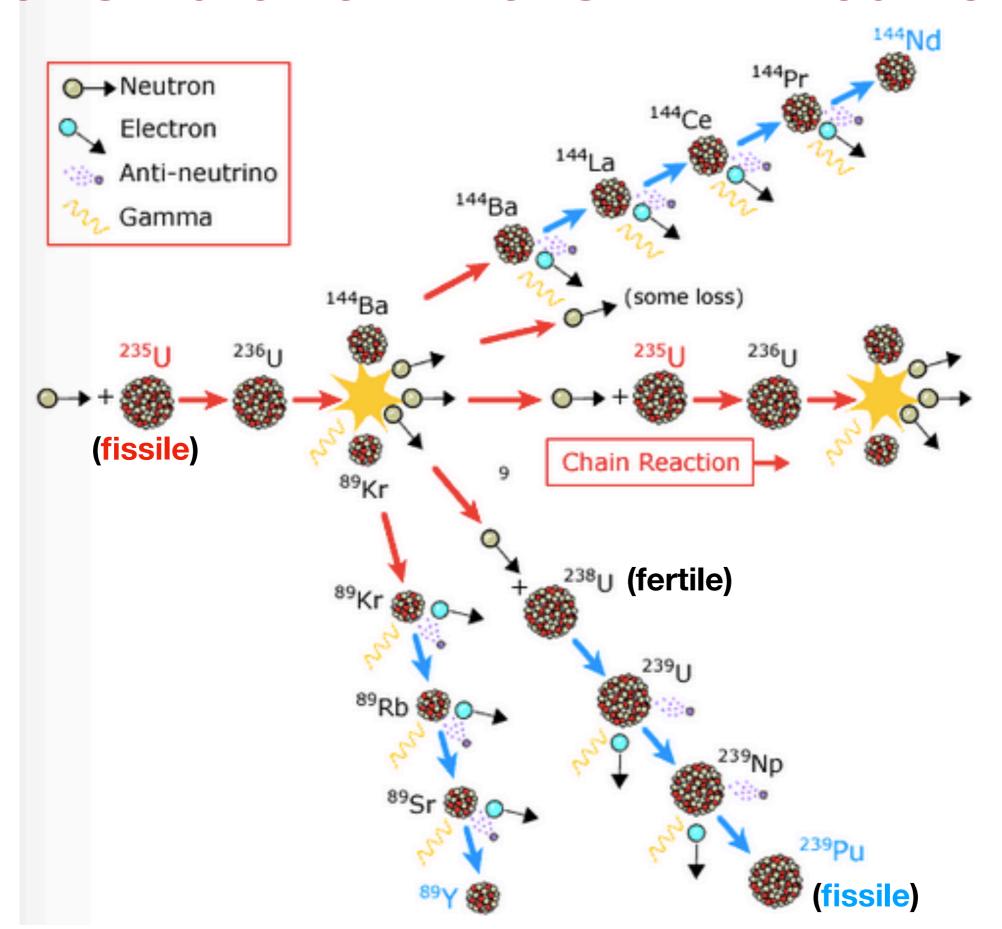
- What's in it?
- What does it do?
- How long does it last?

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- Where should we keep it?

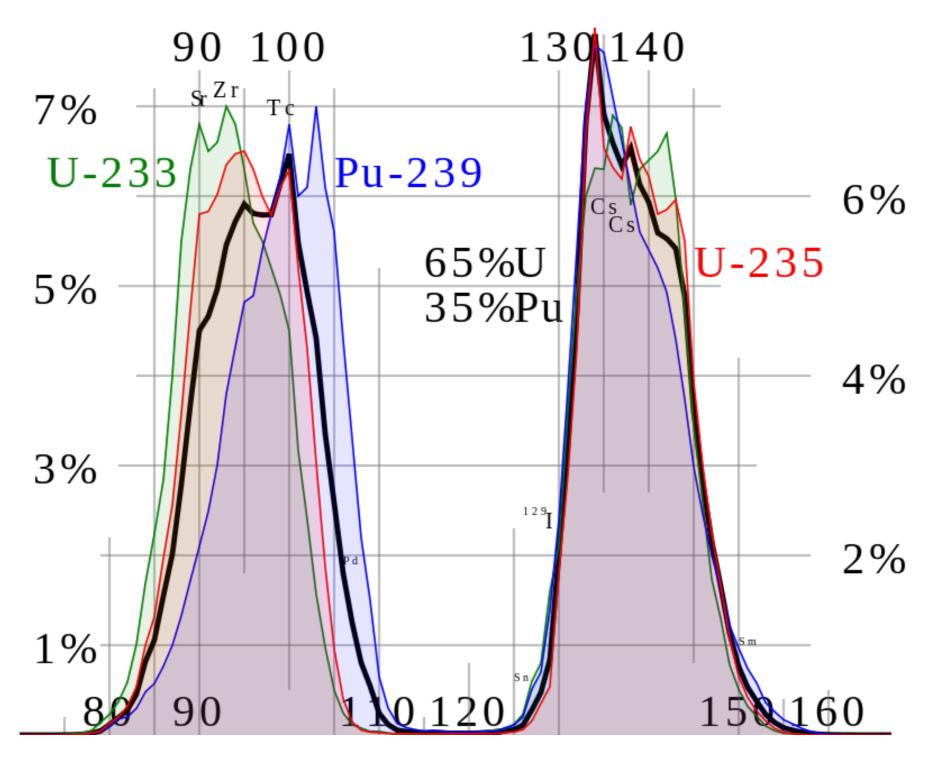
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- What's in it?
- What does it do?
- How long does it last?
- Where should we keep it?
- Is it safe?
- How much does it cost to maintain?

#### **Transmutation tricks with Neutrons:**



## Fission Products



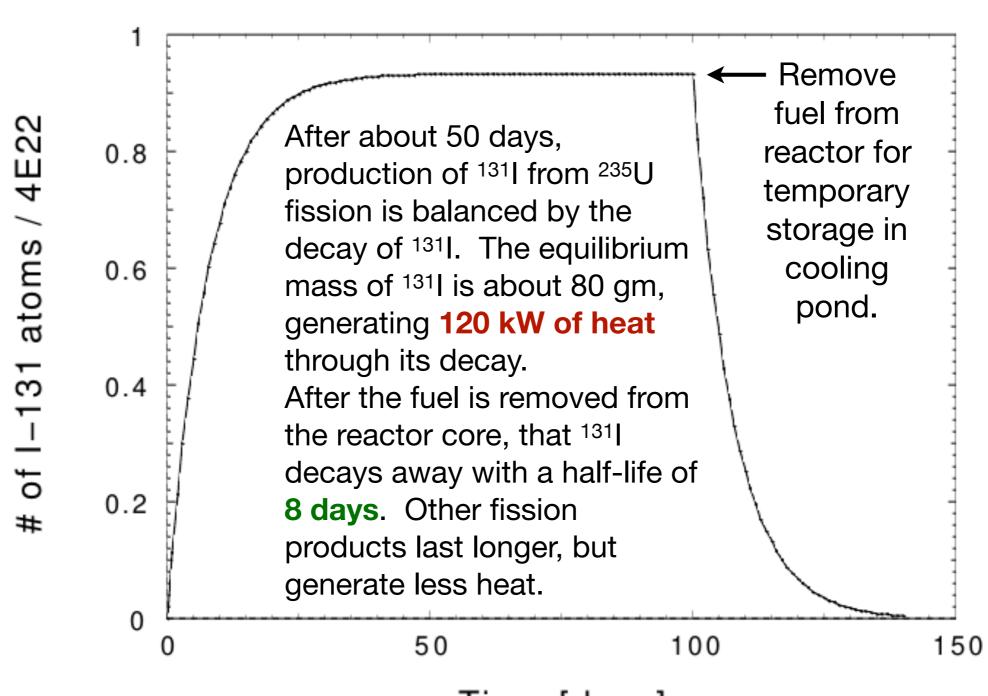
By JWB at en.wikipedia, CC BY 3.0, https://commons.wikimedia.org/w/index.php?curid=16479803

## **Health Concerns**

Isotope	Radiation	Half-life	GI absorption	Notes
Strontium-90/yttrium-90	β	28 years	30%	
Caesium-137	β,γ	30 years	100%	
Promethium-147	β	2.6 years	0.01%	
Cerium-144	β,γ	285 days	0.01%	
Ruthenium-106/rhodium-106	β,γ	1.0 years	0.03%	
Zirconium-95	β,γ	65 days	0.01%	
Strontium-89	β	51 days	30%	
Ruthenium-103	β,γ	39.7 days	0.03%	
Niobium-95	β,γ	35 days	0.01%	
Cerium-141	β,γ	33 days	0.01%	
Barium-140/lanthanum-140	β,γ	12.8 days	5%	
lodine-131	β,γ	8.05 days	100%	
Tritium	β	12.3 years	100%	[a]

# 131 inventory

Buildup & Decay of I-131 in a typical reactor



Time [days]

## Short-Term Storage



Central Interim Storage Facility (CLAB), Sweden. Image: SKB

## **Decay of Fission Products**

Wikipedia: "The radioactivity in the fission product mixture is [initially] mostly *short-lived* isotopes such as <sup>131</sup>I and <sup>140</sup>Ba; after about four months <sup>141</sup>Ce, <sup>95</sup>Zr/<sup>95</sup>Nb and <sup>89</sup>Sr take the largest share, while after about two or three years the largest share is taken by <sup>144</sup>Ce/<sup>144</sup>Pr, <sup>106</sup>Ru/<sup>106</sup>Rh and <sup>147</sup>Pm."

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#### Medium-lived Daughters

```
Prop: t_{1/2} Yield Q * βγ * Unit: (a) (%) (keV)

155Eu 4.76 0.0803 252 βγ

85Kr 10.76 0.2180 687 βγ

113mCd 14.1 0.0008 316 β

90Sr 28.9 4.505 2826 β

137Cs 30.23 6.337 1176 βγ

121mSn 43.9 0.00005 390 βγ

151Sm 88.8 0.5314 77 β
```

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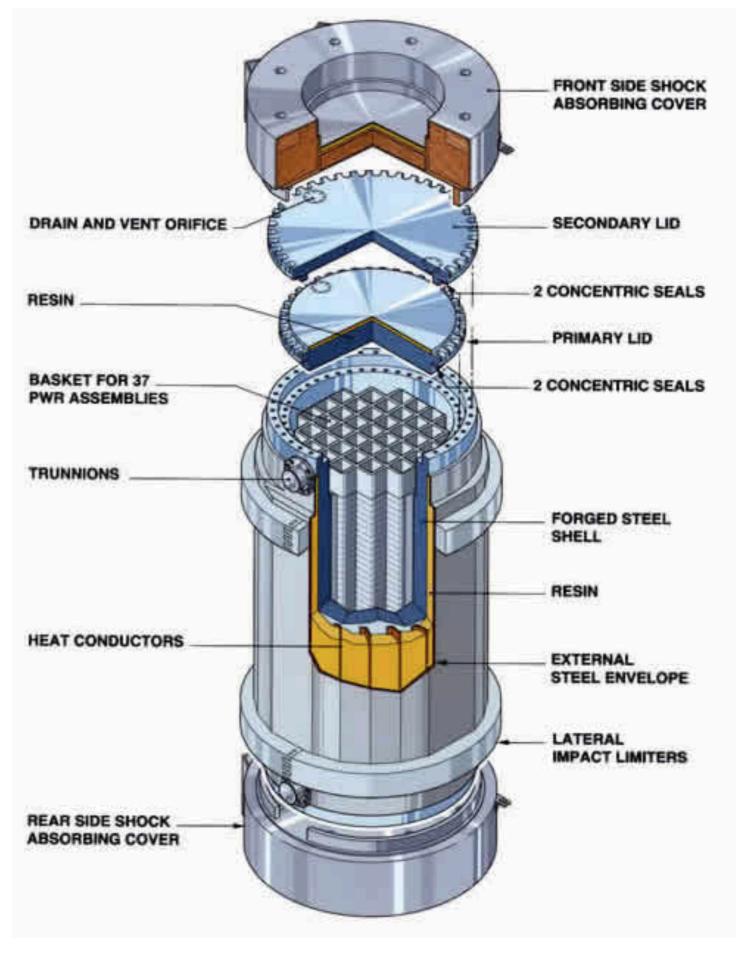
#### **Medium-lived Daughters**

Prop:	t <sub>1/2</sub>	Yield	Q *	βγ *
Unit:	(a)	(%)	(keV)	
<sup>155</sup> Eu	4.76	0.0803	252	βγ
<sup>85</sup> Kr	10.76	0.2180	687	βγ
<sup>113m</sup> Cd	14.1	0.0008	316	β
<sup>90</sup> Sr	28.9	4.505	2826	β
<sup>137</sup> Cs	30.23	6.337	1176	βγ
<sup>121m</sup> Sn	43.9	0.00005	390	βγ
<sup>151</sup> Sm	88.8	0.5314	77	β

#### **Long-lived Daughters**

Nuclide	t1/2	Yield	Decay energy <sup>[a 1]</sup>	Decay mode
<b>\$</b>	(Ma) <b>♦</b>	(%) <sup>[a 2]</sup> ◆	(keV) ♦	<b>♦</b>
<sup>99</sup> Tc	0.211	6.1385	294	β
<sup>126</sup> Sn	0.230	0.1084	4050 <sup>[a 3]</sup>	βγ
<sup>79</sup> Se	0.327	0.0447	151	β
<sup>93</sup> Zr	1.53	5.4575	91	βγ
<sup>135</sup> Cs	2.3	6.9110 <sup>[a 4]</sup>	269	β
<sup>107</sup> Pd	6.5	1.2499	33	β
129	15.7	0.8410	194	βγ

# Accessible Storage ("dry cask")



TN24 cask produced by Orano TN (formerly Areva TN)

# Permanent Storage

## Permanent Storage

- WIPP
- Yucca Mtn
- France
- Sweden
- Finland

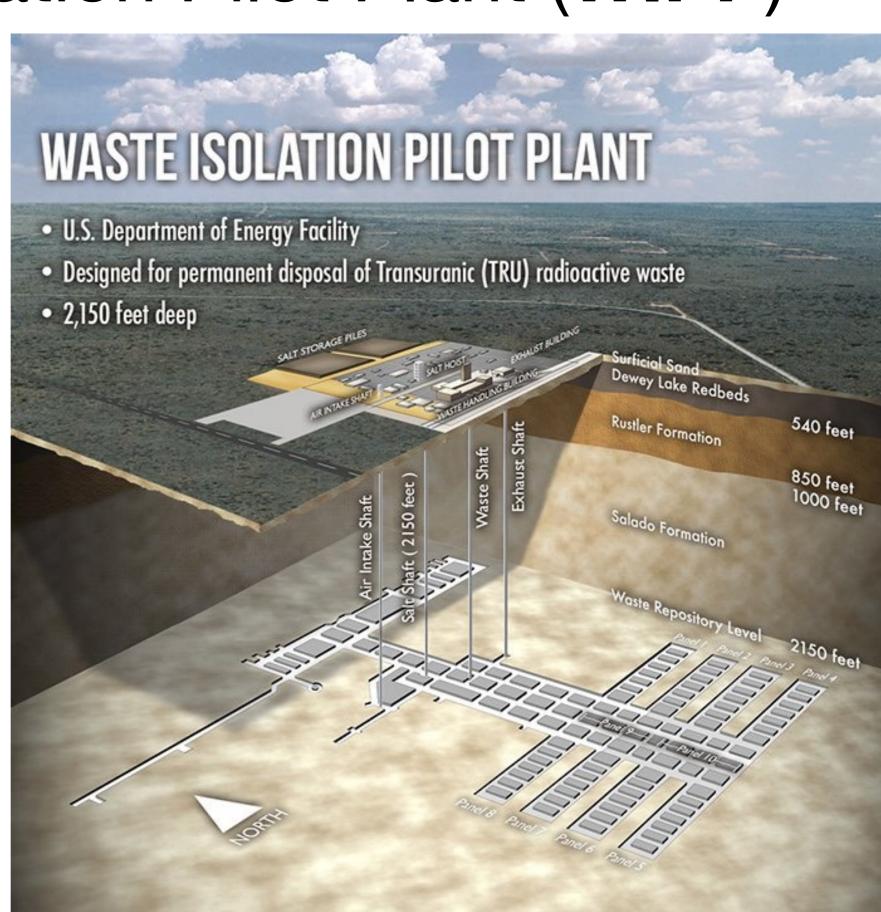
Nuclide	t1/2	Yield	Decay energy <sup>[a 1]</sup>	Decay mode
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## Waste Isolation Pilot Plant (WIPP)

US DOE stores rad.
waste (from nuclear
weapons mfg.) 660 m
underground in a salt
basin near Carlsbad, NM



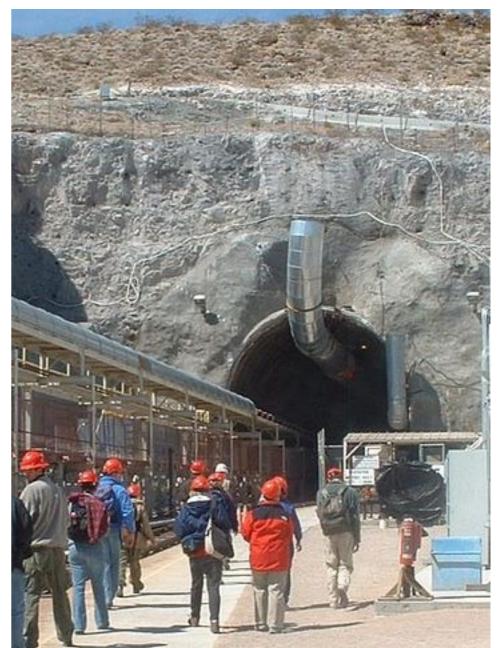
A breach in Feb 2014 released a small amount of radioactivity to the local environment. No one was harmed.



## Yucca Mountain







By Daniel Mayer - Own work, CC BY-SA 3.0, https://commons.wikimedia.org/w/index.php?curid=100567

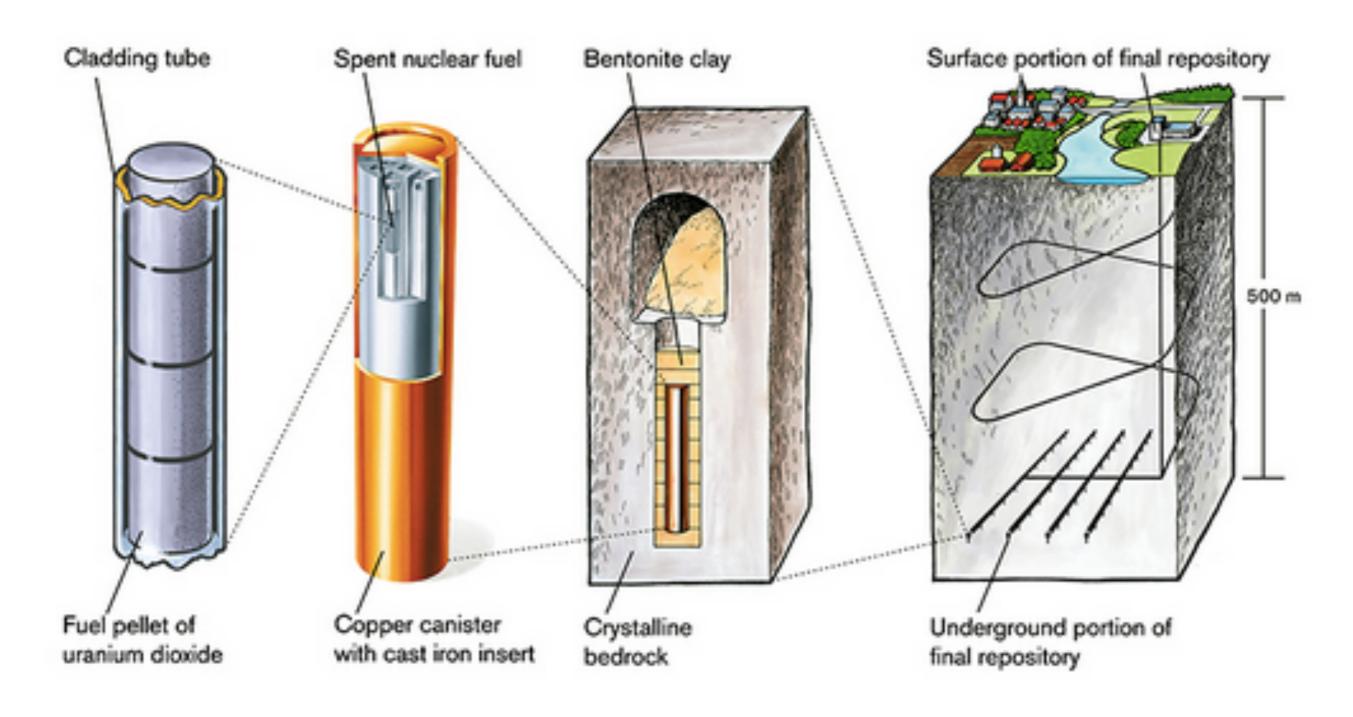
\$US 9 billion project (73% funded by tax on nuclear power) cancelled in 2009 due to opposition by Nevada residents (site is 130 km from Las Vegas).

## Orano La Hague (France)



## SKB: Forsmark, Sweden

KBS-3 method is based on three protective barriers: copper canisters, Bentonite clay and the Swedish bedrock.



## Finland: Olkiluoto Island



## Canada: still thinking about it.

The Canadian Shield offers some of the world's most stable deep bedrock — but why bury something so valuable?

Especially when we can use it as *fuel* in a **M**olten **S**alt **R**eactor!

