# BIOL/PHYS 438 Zoological Physics

- Logistics
- Ch. 4: "Fluids in the Body" wrapup
  - Flow in Pipes
    - Viscosity, Reynolds number & Turbulence
    - Blood Circulation: Aortas to Capillaries
  - Transport of Dissolved Gases

# Logistics

Assignment 1: login, update, Email anytime!

Assignment 2: due Today

#### Assignment 3: due Thursday 15 Feb

Assignment 4: Thu 15 Feb → Tue after Break Spring Break: 17-25 Feb: work on Project too!

## Viscosity $\mu$

A measure of a fluid's resistance to its rate of shear.



### Viscosity vs. Temperature

(Viscosity of water *doubles* from 30°C to 5°C)



# Reynolds Number (Re)

The Reynolds number is the ratio of dynamic pressure  $\rho u^2$  to shearing stress  $\mu u/L$ :

Re = uL/v

- where **U** = velocity of fluid flow (or velocity of object through fluid),
  - L = characteristic length (e.g. diameter of pipe or that of object moving through fluid)
  - and  $V = kinematic viscosity of fluid: <math>v \equiv \mu/\rho$

Flow through a pipe is turbulent for Re > 2300.

## Flow Profile in a Pipe



 $F/A = -\mu du/dr$  locally:  $A = 2\pi rL$  and  $F = \pi r^2 p$  where p is the pressure from the left. Thus  $du/dr = -(p/2\mu L)r$ , which gives

 $u(r) = (p/4\mu L)[R^2 - r^2]$ 

When the Reynolds number *Re* exceeds about 2300, the flow becomes *turbulent*.

#### Average Flow Velocity in a Pipe



#### Doh! Du Jour



What's wrong with this picture?



### Laminar *vs.* Turbulent Flow

Photo by Friedrich Ahlborn [1918]





# Pipe Resistance



### Laminal Flow Control



### The Aorta



 $\begin{aligned} R_{\text{aorta}} & [\text{m}] \ge 1.2 \cdot 10^{-4} M^{3/4} \\ A_{\text{aorta}} & [\text{m}^2] \ge 4.5 \cdot 10^{-8} M^{3/2} \\ u_{\text{aorta}} & [\text{m/s}] \le 31.6 M^{-3/4} \end{aligned}$ 

Circuits



# Lungs & Alveoli



#### Hæmoglobin

 $\mu = -\tau \, d\sigma/dN$  is like a potential energy [J]: oxygen molecules tend to move "downhill" from high  $\mu$  to low  $\mu$ . For concentrations of solutes in water we have  $\mu_{IG} = \tau \log(n/n_Q)$  where  $n_Q$  is a constant. In thermal equilibrium, we require  $\mu_{tot} = \mu_{IG} + \mu_{ext} = \text{constant}$ , where  $\mu_{ext}$  is the binding energy of an  $O_2$  molecule to hæmoglobin (Hb). The stronger the binding, the more "downhill"! The density *n* is proportional to the partial pressure *p*. Oxygen occupies all 4 Hb sites for p > 10 kPa (~0.1 atm) and is released when  $p < 2 \, \text{kPa} (\sim 0.02 \, \text{atm})$ . What happens when  $CO_2$  competes with  $O_2$  for Hb sites?

#### Aorta to Capillaries and Back



#### **Heart Specs**



 $P_{\text{heart}} [W] \approx 1.95 \cdot 10^{-2} \ b \ M^{34} = f_{\text{heart}} \Delta V_{\text{heart}} \ \Delta p_{\text{heart}}$  $\Delta V_{\text{heart}} \sim M \quad \text{and} \quad \Delta p_{\text{heart}} \text{ is independent of } M$ so  $f_{\text{heart}} \sim M^{-34} \qquad [\text{man: } \sim 1 \ \text{Hz; mouse: } \sim 9.2 \ \text{Hz}]$