## BIOL/PHYS 438

## Zociogioal Physics

## Logistics

Assignment 1: login, update, Email anytime!
Assignment 2: due Today
Assignment 3: due Thursday 15 Feb
Assignment 4: Thu $15 \mathrm{Feb} \rightarrow$ Tue after Break
Spring Break: 17-25 Feb: work on Project too!

## Viscosity vs. Temperature

A measure of a fluid's resistance to its rate of shear.

(Viscosity of water doubles from $30^{\circ} \mathrm{C}$ to $5^{\circ} \mathrm{C}$ )


## Reynolds Number (Re)

The Reynolds number is the ratio of dynamic pressure $\rho u^{2}$ to shearing stress $\mu u / L$ :

$$
R e=u L / v
$$

where $U=$ velocity of fluid flow (or velocity of object through fluid),
$L=$ characteristic length (e.g. diameter of pipe or that of object moving through fluid)
and $\quad V=$ kinematic viscosity of fluid: $V \equiv \mu / \rho$
Flow through a pipe is turbulent for $R e>2300$

## Average Flow Velocity in a Pipe



The area-weighted average of

$$
\begin{gathered}
u(r)=(\Delta p / 4 \mu L)\left[R^{2}-r^{2}\right] \text { is } \\
u_{\mathrm{av}}=\Delta p R^{2} / 8 \mu L
\end{gathered}
$$

and the mass flow rate $J$ is

$$
J=\rho u_{\mathrm{av}} \pi R^{2}=\frac{\pi R^{4}}{8 v} \cdot \frac{\Delta p}{L}
$$

Hagen Poiseuille pipe resistance
$\lambda_{\mathrm{HP}}=8 \mathrm{~V} / \pi R^{4}$

$$
\lambda_{\mathrm{HP}}=8 \mathrm{VL} / \pi R
$$

$\qquad$ - $J=\Delta p / \lambda_{\mathrm{HP}}$

Flow Profile in a Pipe

$F / A=-\mu \mathrm{d} u / \mathrm{d} r$ locally: $A=2 \pi r L$ and $F=\pi r^{2} p$ where $p$ is the pressure from the left. Thus $\mathrm{d} u / \mathrm{d} r=-(p / 2 \mu \mathrm{~L}) r$, which gives

$$
u(r)=(p / 4 \mu L)\left[R^{2}-r^{2}\right]
$$

When the Reynolds number Re exceeds about 2300, the flow becomes turbulent.

Doh! Du Jour


What's wrong with this picture?


## Laminar vs. Turbulent Flow



Pipe Resistance


## Laminal Flow Control

The Aorta



## Harmoglobin

$\mu=-\tau \mathrm{d} \sigma / \mathrm{d} N$ is like a potential energy [J]: oxygen molecules tend to move "downhill" from high $\mu$ to low $\mu$. For concentrations of solutes in water we have $\mu_{\mathrm{IG}}=\tau \log \left(n / n_{\mathrm{Q}}\right)$ where $n_{\mathrm{Q}}$ is a constant. In thermal equilibrium, we require $\mu_{\mathrm{tot}}=\mu_{\mathrm{IG}}+\mu_{\mathrm{ext}}=$ constant, where $\mu_{\text {ext }}$ is the binding energy of an $O_{2}$ molecule to hœemoglobin $(\mathrm{Hb})$. The stronger the binding, the more "downhill"! The density $n$ is proportional to the partial pressure $p$. Oxygen occupies all 4 Hb sites for $p>10$ $\mathrm{kPa}(\sim 0.1 \mathrm{~atm})$ and is released when $p<2 \mathrm{kPa}(\sim 0.02$ atm). What happens when $\mathrm{CO}_{2}$ competes with $\mathrm{O}_{2}$ for Hb sites?

## Aorta to Capillaries and Back



## Heart Specs



